

# QUI-42070: Termodinámica Avanzada

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## ***Programa QUI-42070***

### ***Información general***

**HORARIO:** 10h-12h Martes y Jueves

**LOCAL:** Aula Zamman

**PROFESOR:** Dr. Roger Coziol (rcoziol@astro.ugto.mx)

**OFICINA:** Departamento de Astronomía, en valenciana (al lado del CIMAT)  
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**Notas de cursos:** <http://www.astro.ugto.mx/cursos/qui42070.htm>

### ***Exámenes, Tareas y Evaluación***

**Tareas:** aprox. una cada dos semanas

**Exámenes:** 2 exámenes parciales

#### **Clasificación final:**

Exámenes Parciales ..... 60%

Tareas.....40%

## *Bibliografia*

1. Sethna J. P., 2006:  
Statistical Mechanics:  
Entropy, Order Parameters, and Complexity  
Oxford University Press
2. **Dill, K. A. & Bromberg, S., 2003:**  
**Molecular Driving Forces:**  
**Statistical Thermodynamics in Chemistry and Biology**  
**Garlan Science**
3. McQuarrie D. A., 2000:  
Statistical Mechanics  
University Science Books
4. Sears, F. W. & Salinger, G. L., 1986, 3<sup>rd</sup> ed.:  
Thermodynamics, Kinetic Theory,  
And Statistical Thermodynamics  
Addison-Wesley
5. Tolman R. C. 1979:  
The Principles of Statistical Mechanics  
Dover
6. Kittel, C., 1969:  
Thermal Physics  
John Wiley & Sons
7. Yourgrau, W., van der Merwe, A. & Raw G. 1966:  
Treatise on Irreversible and statistical Thermophysics  
The MacMillan Company
8. Reif, F., 1965:  
Fundamentals of Statistical and Thermal Physics  
McGraw-Hill

## Temario Termodinámica Avanzada

<p>1. Introduction</p> <ul style="list-style-type: none"> <li>• Extremum principle</li> <li>• Heat work and energy</li> <li>• Entropy and the Boltzmann Distribution law</li> </ul>	<p>2. Thermodynamics</p> <ul style="list-style-type: none"> <li>• Thermodynamics driving forces</li> <li>• Free energies</li> <li>• Maxwell's relations &amp; mixtures</li> </ul>
<p>3. Statistical mechanics</p> <p>Part 1</p> <ul style="list-style-type: none"> <li>• Boltzmann Distribution law</li> <li>• Statistical mechanics of simple gases and solids</li> </ul> <p>Part 2</p> <ul style="list-style-type: none"> <li>• Temperature, heat capacity</li> <li>• Chemical equilibrium</li> <li>• Equilibria between liquids, solids and gases</li> </ul> <p>Part 3</p> <ul style="list-style-type: none"> <li>• Solutions and mixtures</li> <li>• Solvation and transfers of molecules between phases</li> </ul>	<p>4. Kinetics and electromagnetism</p> <p>Part 1</p> <ul style="list-style-type: none"> <li>• Physical Kinetics</li> <li>• Chemical Kinetics and transition status</li> </ul> <p>Part 2</p> <ul style="list-style-type: none"> <li>• Coulomb's law and electrostatic potential</li> <li>• Electrochemical equilibria</li> </ul> <p>Part 3</p> <ul style="list-style-type: none"> <li>• Electrolytes</li> <li>• Molecular interactions</li> </ul>
<p>5. Phase transition and binding processes</p> <p>Part 1</p> <ul style="list-style-type: none"> <li>• Phases transitions</li> <li>• Cooperativity</li> </ul> <p>Part 2</p> <ul style="list-style-type: none"> <li>• Adsorption, Binding and Catalysis</li> <li>• Multi-site Cooperative Ligand Binding</li> </ul>	<p>6. Water</p> <ul style="list-style-type: none"> <li>• Unusual characteristics</li> <li>• Water as a solvent</li> </ul>
<p>7. Polymers</p> <ul style="list-style-type: none"> <li>• Polymer solutions</li> <li>• Polymer elasticity</li> <li>• Polymer confinement and deformations</li> </ul>	<p>8. Extra</p> <ul style="list-style-type: none"> <li>• Principles of probability</li> <li>• Series and approximations</li> <li>• Multivariate calculus</li> <li>• Legendre transform</li> <li>• Vector calculus</li> </ul>